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The Photochemical Synthesis of 6-Acyl-5*H*-benzo[a]phenoxazin-5-ones

Short Communication

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6-Acyl-5H-benzo[a]phenoxazin-5-ones were prepared by the photochemical reaction of 5H-benzo[a]phenoxazin-5-one with aldehydes.

(Keywords: 5H-Benzo[a]phenoxazin-5-one; 6-Propionyl-5H-benzo[a]-phenoxazin-5-one)

Die photochemische Synthese von 6-Acyl-5H-benzo[a]phenoxazin-5-onen (Kurze Mitteilung)

Die Titelverbindungen wurden mittels photochemischer Reaktion von 5H-Benzo[a]phenoxazin-5-onen mit Aldehyden hergestellt.

Although many workers have reported synthetic methods of a variety of benzophenoxazones which are known to possess pharmacological activities¹⁻³, the recorded synthetic methods are tedious⁴⁻⁸. On the other hand, photochemistry has provided versatile routes to many compounds which are difficult to be prepared by any other methods. We now wish to report a novel route for the photochemical synthesis of 5H-benzo[a]phenoxazin-5-one derivatives.

Upon irradiation of benzene solution of 5H-benzo[a]phenoxazin-5-one (1) and aldehydes (2a, b) in a Pyrex vessel using 100 Watt high-pressure mercury lamp, 6-acyl-5H-benzo[a]phenoxazin-5-ones (3a, b) were obtained. The structures of 3a and 3b were fully supported by microanalytical results and spectral data. In particular, the NMR spectrum of 1 (CDCl₃) exhibited a characteristic singlet at 6.30 ppm due to the olefinic proton, but those of the photoproducts 3a, b of 1 with 2a

a: R = CH3

b: R = C₂H₅

and 2 b did not show a characteristic singlet at 6.30 ppm. Therefore, the photochemical addition did occur at the 6-position of 1. Furthermore, the photoproduct 3 a was identified by direct comparison with a sample prepared by an alternate route. This route involves the condensation of 2-aminophenol (4) with 2-acetyl-3-hydroxy-1,4-naphthoquinone (5) in an aqueous acetic acid. The latter compound was obtained in nine steps from 1,4-naphthoquinone⁹.

The present report offers a facile method for the photochemical synthesis of 6-acyl-5 H-benzo[a]phenoxazin-5-ones from readily available starting material.

Experimental

Melting points were determined on a Yanagimoto micromelting apparatus and are uncorrected. The infrared spectra were recorded on a Jasco DS 701 G spectrometer. Nuclear magnetic resonance spectra were determined on a Hitachi R-20 B spectrometer using tetramethylsilane as an internal reference. Mass spectra were determined on a Hitachi M-52 spectrometer.

6-Acetyl-5H-benzo[a]phenoxazin-5-one (3 a) Method (A)

A benzene solution (50 ml) of 1 (0.005 mol) and 2a (0.03 mol) was irradiated in a Pyrex glass tube for 20 h from outside by means of 100 Watt high-pressure mercury arc lamp in a nitrogen atmosphere through a 7 cm-thick layer of cold water (0-5°) and then bubbled with air for several minutes. After the removal of the solvent under reduced pressure, the residue was chromatographed on a silica gel column using cyclohexane-benzene (1:1) as an eluent. The photoproduct 3a thus obtained was further purified by recrystallization from ethanol.

Compound 3 a gave yellow crystals; M. p. 203°; yield 52%.

IR (KBr): $1618 \,\mathrm{cm}^{-1}$ (C=O).

 $^1H\text{-NMR}$ (CDCl₃): 2.60 (s, 3 H, CH₃ group), 7.09–7.85 (m, 6 H, arom.), 8.07–8.29 (m, 1 H, arom.), 8.47–8.67 (m, 1 H, arom.).

MS: $m/s = 289 (M^+)$.

 $C_{18}H_{11}O_3N$ (289.3). Calcd. C 74.73, H 3.83, N 4.84. Found C 74.75, H 3.72, N 4.85.

Method (B)

A mixture of 4 (0.0004 mol) and 5 (0.0009 mol) in 5 ml of 90% aqueous acetic acid was heated at 100 °C for 2 h. After the removal of the solvent under reduced pressure, the residue was chromatographed on a silica gel column using cyclohexane-benzene (1:1) as an eluent. The crude product was further purified by recrystallization from ethanol and identified (M. p., mixed M. p., IR, 1 H-NMR, and MS) as 3a. Yield 45%.

6-Propionyl-5H-benzo[a]phenoxazin-5-one (3b)

The compound 3b was prepared by photochemical reaction of 1 with 2b in analogy to 3a.

Compound 3b gave yellow crystals; M. p. 177°; yield 62%.

IR (KBr): $1619 \,\mathrm{cm}^{-1}$ (C=O).

 $^1\mathrm{H\text{-}NMR}$ (CDCl₃): 0.99-1.33 (t, 3 H, CH₃ group), 2.49-2.94 (q, 2 H, CH₂ group), 6.96-7.78 (m, 6 H, arom.), 8.01-8.30 (m, 1 H, arom.), 8.41-8.68 (m, 1 H, arom.).

 $MS: m/e = 303 \ (M^{+}).$

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